Additional Information

Remember, the type of water you have will determine what type of treatment is possible. No one treatment technique works for every iron problem and well

construction or reconstruction may be more cost effective. For additional information on iron in drinking water, contact a water treatment dealer, a licensed plumber or a licensed well driller or pump installer.

Table 1:	Treatment	Consideration	ns for Variou	ıs Forms of Iro	n
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Characteristic	Known As	Treatment Methods	Considerations
Drawn tap water is clear and colorless.	Soluble Clear Water Fe+2 Ferrous Dissolved	Aeration/Filtration	May require lengthy contact time. Temperature dependent.
hen allowed to and, reddish brown articles appear and		Water softener	Hardness must be calculated. System must be airtight. All water must be treated.
settle to bottom.		Chlorination/Filtration	Chlorine liquid or pellets. Frequent monitoring. Proper water pressure.
		Manganese greensand/Filtration	Adequate pressure.
		Catalytic filtration "BIRM"	Dissolved oxygen, organic matter, chlorination, polyphosphate, temperature limitations
		Ozonation	Used by some municipal systems. Expense
		Sequestering	May not prevent staining. May need to remove sequestering agents and iron. Test for agents before choosing another treatment device.
Drawn tap water	Insoluble Red Water Fe+3 Ferric Oxidized	Manganese greensand/Filtration	Adequate pressure.
appears rusty or has a red or yellow color. When allowed		Catalytic filtration "BIRM"	Dissolved oxygen, alkalinity, organic matter, chlorination, polyphosphate, temperature limitations
o stand, particles ettle to bottom.		Chlorination/Filtration	Chlorine liquid or pellets. Frequent monitoring. Proper water pressure.
Water tank/toilet tank/plumbing have reddish brown or yellow gelatinous slime or sludge present. May have objectionable odor or oily sheen.	Bacterial Creno-thrix Leptothrix Gallionella	Shock chlorination and consider following with continuous chlorination. Bactericides.	Shock chlorination should include: cleaning the well thoroughly, cleaning pump and riser pipe, and complete chlorination and flushing of distribution system. Make sure bactericides can be used in drinking water. Bactericides need long contact time for adequate treatment.
High color content (yellow or brown) or colorless. Generally	Organic Hemme Tannin	Water softener	First step is to treat for organics. Hardness must be calculated. System must be airtight. Treat all water.
groundwater from shallow well or		Manganese greensand/Filtration	First step is to treat for organics. Adequate pressure.
surface water.		Ozonation	Used by some municipal systems. Expense

Department of Natural Resources Offices



Northern Region

810 W. Maple Street Spooner, WI 54801 (715) 635-2101

107 Sutliff Avenue Rhinelander, WI 54501 (715) 365-8900

South Central Region

3911 Fish Hatchery Rd. Fitchburg, WI 53711 (608) 275-3266

(715) 839-3700

West Central Region 1300 W. Clairemont PO Box 4001 Eau Claire, WI 54702-4001

Southeast Region

2300 N. Dr. Martin Luther King, Jr. Drive Milwaukee, WI 53212 (414) 263-8500

Northeast Region

2984 Shawano Avenue P.O. Box 10448 Green Bay, WI 54307-0448 920-662-5100

Central Office

101 S. Webster P.O. Box 7921 Madison, WI 53707-7921 (608) 266-0821



Introduction

ron is one of the earth's most plentiful resources, **_** making up at least five percent of the earth's crust. When rainfall seeps through the soil, the iron in the earth's surface dissolves, causing it to go into almost every natural water supply, including well water. When iron is present in our water, it is usually found at concentrations less than 10 milligrams per liter (mg/l) or parts per million (ppm); however, higher levels are often found.

This brochure was revised by the Wisconsin Department of Natural Resources with assistance from the Education Subcommittee of the Groundwater Coordinating Council.

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This publication is available in alternative format (large print, Braille, audio tape, etc) upon request. Please call (608) 266-0821 for more information.



Bureau of Drinking Water & Groundwater Wisconsin Department of Natural Resources

Health and Water Quality

Iron is not considered hazardous to health. In fact, iron is essential for good health because it transports oxygen in your blood. In the United States, most tap water probably supplies less than 5 percent of the dietary requirement for iron.

Under Department of Natural Resources (DNR) rules, iron is considered a secondary or "aesthetic" contaminant. The present recommended limit for iron in water, 0.3 mg/l (ppm), is based on taste and appearance rather than on any detrimental health effect. Private water supplies are not subject to the rules, but the guidelines can be used to evaluate water quality.

For instance, when the level of iron in water exceeds the 0.3 mg/l limit, we experience red, brown, or yellow staining of laundry, glassware, dishes, and household fixtures such as bathtubs and sinks. The water may also have a metallic taste and an offensive odor. Water system piping and fixtures can also become restricted or clogged.

Types of Iron

Iron is generally divided into two main categories: 1) soluble or ferrous and 2) insoluble or ferric iron. Soluble iron, or "clear water" iron, is the type of iron found in our groundwater and oxidizes to insoluble or red iron in the presence of oxygen either in the well or in your home. This type of iron is identified after you've poured a glass of cold clear water. If allowed to stand in the presence of air, reddish brown particles will appear in the glass and eventually settle to the bottom.

When insoluble iron, or "red water" iron is poured into a glass, it appears rusty or has a red or yellow color. Insoluble iron can create serious taste and appearance problems for the water user.

Iron, which combines with different naturallyoccurring organic acids or tannins, may also exist as an organic complex. A combination of acid and iron, or **organic iron**, can be found anywhere; however, it is more common in shallow wells and surface water. Although this kind of iron can be colorless, it is usually yellow or brown.

Finally, when iron exists along with certain kinds of bacteria, problems can become even worse. The bacteria consume iron to survive and leave a reddish brown or yellow slime that can clog plumbing and cause an offensive odor. You may notice this slime or sludge in your toilet tank when you remove the lid. For more information on iron bacteria, find the DNR publication, Iron Bacteria Problems in Wells on the internet at dnr.wi.gov/org/water/dwg/febact.htm. Once you determine whether you have "clear water," "red water," "organic" or "bacterial" iron in your water, you can take steps to correct the problem. Keep in mind that no one treatment method will work for every type of iron problem.

Test Your Water

Before you attempt to remove anything that appears to be iron-related, it is important to have your water tested. A complete water test to determine the extent of your iron problem and possible treatment solutions should include tests for iron concentration, iron bacteria, pH, alkalinity, and hardness. Check the business pages of the phone book under "Laboratories-Testing" to find a water testing lab or call a licensed professional for assistance.

If you receive your water from a public water system and experience red water problems, it is important to contact a utility official to determine whether the red water is from the public system or your home's plumbing or piping.

Well Construction/Reconstruction

High iron levels may be avoided in some cases by changing the screen or casing depth of the well as long as the minimum casing depth requirements are still met. Talking to your neighbors about their well depths and iron levels will give you some idea of what well depth would pump the lowest amount of iron. It is also helpful to talk to a well driller or pump installer about local conditions and the cost of drilling a new well in your area. The cost of well work should be compared to the long term (perhaps twenty years) cost of treating the water for any iron related problems.

Iron Treatment Terms

Aeration: Introducing oxygen to the water source to convert soluble iron to its insoluble form.

Filtration: Media used to entrap and screen out oxidized particles of iron. Usually requires backwashing to remove accumulated iron.

Water Softening: Removal of soluble iron by ion exchange.

Manganese Greensand: An ion exchange sand material which is capable of removing iron. Adsorbs dissolved iron and requires chemical regeneration.

Catalytic Filtration "BIRM": A granular filter medium that enhances the reaction between oxygen and iron and then filters the insoluble iron.

Ozonation: A specialized form of aeration using ozone to convert soluble iron to insoluble iron.

Ion Exchange: Substituting an acceptable ion (such as sodium) for soluble iron.

Sequestering: Adding chemical agents to water to keep metals like iron in solution to prevent characteristic red stains.

Chlorination: Chemical oxidizer used to convert soluble iron to an insoluble, filterable form.





Treatment

Table 1 lists treatment considerations for the various forms of iron. For additional information on water treatment systems, contact your County Extension Office or Extension Publications, Rm. 245, 30 N. Murray Street, Madison, WI 53715 and ask for publication G3558-5, Choosing a Water Treatment Device or you could contact a licensed plumber.

When choosing a water treatment method or device, make sure you have answers to the following five questions:

- 1. What form of iron do I have in my water system?
- Will the water treatment unit remove the total iron concentration (determined by the water test) in my water supply? (Total iron refers to both soluble and insoluble iron combined).
- 3. Will the treatment unit treat the water at the flow rate required for my water system?
- 4. Considering the results of my water test, will this method effectively remove iron? (For example, pH may need to be adjusted before beginning a particular treatment).
- 5. Would well construction or reconstruction be more cost effective than a long term iron removal treatment process?